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constant, as 0.0821 l atm
 $\text{K}^{-1} \text{ mol}^{-1}$. a.

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1B.1(b) The mean speed is

$8RT$ v_{mean} M The mean

translational kinetic energy

is m $3RT$ $3kT$ 2 M 2 The

ratios of species 1 to

species 2 at the same

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temperature are E_k , $v_{\text{mean},1}$

$v_{\text{mean},2}$ (i) $\frac{1}{2} m v_1^2 = \frac{1}{2} m v_2^2$

$v_{\text{mean},H} = \sqrt{\frac{2E_k}{m}}$ and $v_{\text{mean},Hg} = \sqrt{\frac{2E_k}{m}}$

$v_{\text{mean},Hg} = \sqrt{\frac{2 \times 7.079}{200.6}} = 0.267 \text{ m/s}$

$v_{\text{rms}} = \sqrt{\frac{3E_k}{m}}$ (ii)

The mean translation kinetic energy is independent of molecular mass and depends

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upon temperature alone!

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